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# Plant-extract-mediated biosynthesis of silver nanoparticles using Eleutherine americana bulb extract and its characterization

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**Abstract.** *Manullang JR, Nugroho RA, Rohmah M, Rudianto, Qorysuchi A. 2021. Plant-extract-mediated biosynthesis of silver nanoparticles using* Eleutherine americana *bulb extract and its characterization. Nusantara Bioscience 13: 247-254.* The plant-based biosynthesis of nanoparticles has gained increasing momentum due to being lower in cost and eco-friendly. This study aimed to biosynthesize nanoparticles from the bulb ethanolic extract of *Eleutherine americana* (Aubl.) Merr. ex K.Heyne (Ea-AgNPs), then characterize Ea-AgNPs and determine their phytochemical content and antioxidant capacity. The Ea-AgNPs were synthesized using ethanolic extract of *E. americana* bulb along with various concentrations of AgNO<sub>3</sub> (0.5-4 mM). The Ea-AgNPs were then characterized using UV-VIS spectroscopic, Scanning Electron Microscopy/ Energy Dispersive X-ray (SEM/EDX), Transmission Electron Microscopy (TEM), X-ray Powder Diffractometry (XRD), and Fourier-Transform Infrared Spectroscopy (FTIR) techniques. The results indicated that *E. americana* can be used to reduce AgNO<sub>3</sub> to synthesize Ea-AgNPs, indicated by color change, and had optimum UV/VIS spectra at 400 nm. The FTIR analysis found that Ea-AgNPs showed peaks at 2919, 2850, 1586, and 1031 cm<sup>1</sup>, containing several important biocompounds. Additionally, the XRD results found an amorphous Ea-AgNP peak with maximum intensity and proportion of silver occurring at 24 Theta. The particle size distribution curve of Ea-AgNPs showed a size of 10<sup>5</sup> nm. Furthermore, SEM/EDX analysis revealed an optical absorption characteristic peak at 3 keV. The EDX examination revealed three signals: a strong signal from the C atom (70.99 %), an O atom (28.95 %), and Ag atom (0.06 %). The TEM imaging also showed the characteristics of Ea-AgNPs. Some phytochemicals such as flavonoids, tannins, alkaloids, and saponins were found in Ea-AgNPs, with IC<sub>50</sub> values of 45.30 ppm.

Keywords: Antioxidant properties, characterization, Eleutherine americana, nanoparticle

# INTRODUCTION

Medicinal plant extracts were used in a bottom-up single pot synthesis strategy using a wet chemistry method to conduct green synthesis of silver nanoparticles. This method is environmentally benign, low-nanotechnology has arisen from many disciplines of science and engineering, where novel concepts for modifying molecules and single atoms have been developed (Usman et al. 2020; Gottardo et al. 2021; Zhang et al. 2021). The application of nanomaterials is particularly popular within environmental and medical research in the fast-developing field of nanotechnology. Silver nanoparticles are distinctive among metallic nanoparticles (Crisan et al. 2021; Kaabipour and Hemmati 2021; Restrepo and Villa 2021); they are the most widely investigated by researchers worldwide due to their versatile applications, simplicity of synthesis, adaptability, shape, and large surface area.

Nanoparticles are also very broadly applied in fields such as biomedicine, agriculture, and poultry feed. In biomedicine, nanoparticle plant-based extract demonstrated significant antibacterial activity against a variety of pathogenic microorganisms, in addition to cytotoxic action in the A549 and HepG2 cell lines (Acay 2021). Meanwhile, several studies have investigated the effects of nanoparticles on seed germination, growth stimulation, and metabolic rate changes (Goswami and Mathur 2019). However, nanoparticles also have negative consequences such as suppressing plant development, inhibiting chlorophyll synthesis, and lowering photosynthetic efficiency (Goswami and Mathur 2019). In poultry, because of their huge surface area to volume ratio and rapid absorption in the body, nanoparticles may be introduced to animal feed, providing an ideal platform for incorporating diverse range of substances such as vaccinations and vitamin supplements. Nanoparticles can also facilitate the direct delivery of chemicals to specific organs or systems while avoiding the rapid degradation seen with antibiotics, resulting in a variety of health advantages (Singh and Lillard 2009).

Nanoparticles are often made using a range of chemical and physical processes that are both costly and potentially harmful to the environment (Guo 2012; Uzair et al. 2020). As a result, green synthesis approaches are gradually being linked with current scientific and industrial advancements in the global endeavor to decrease the production of

hazardous waste. In this study, cost, and produces stable, well-dispersed silver nanoparticles with little aggregation and good size control (Iravani and Varma 2020; Velsankar et al. 2020).

In the present study, an extract of Eleutherine americana (Aubl.) Merr. ex K.Heyne was employed as a reducing and stabilizing agent. Eleutherine americana (Dayak onion) is traditionally used as an ethnomedicine (Kamarudin et al. 2021); the bulb of this species is an essential traditional medicine used by Davak tribes to treat heart illness, as an anti-inflammatory, to improve breastproduction, diabetes, breast cancer, stroke, hypertension, sexual problems, and as a bacterial infection therapy (Ifesan et al. 2009; Song et al. 2009; Saragih et al. 2014; Nuryanto and Paramita 2018). The bulbs contain naphthalene, naphthoquinone, and anthraquinone (Mahabusarakam et al. 2010; Insanu et al. 2014), in addition to phenolic compounds, which have powerful antioxidant effects and may be involved in the treatment of some diseases (Kuntorini et al. 2016). Furthermore, an oligosaccharide extract from E. americana has been utilized as a prebiotic to promote the development of gut flora (Phoem et al. 2019).

Plant extract as a reducing and stabilizing agent in the green production of silver nanoparticles is a new approach that is of interest to numerous scientists. Fatimah and Aftrid (2019) worked on biosynthesis of AgNPs using red spinach (Amaranthus tricolor) leaf extract and investigated its antibacterial activity, while Kedi et al. (2018) synthesized AgNPs from Selaginella myosurus. Another study performed by Jalilian et al. (2020) focused on the green synthesis of AgNPs using Allium ampeloprasum aqueous extract and investigated their characterization, antioxidant activity, and antibacterial and cytotoxicity effects. Vijayakumar et al. (2019) used garlic clove extract to produce AgNPs and further assessed their antibacterial, antibiofilm, anthelminthic, anti-inflammatory, anticancer, and ecotoxicity characteristics. Other research on the biosynthesis of AgNPs using various plant extracts are as follows: Oscillatoria sp. extracts (Adebayo-Tayo et al. 2019), cauliflower extract (Oda et al. 2019), and the bulb of Myrmecodia sp. (Nugroho et al. 2021).

Despite the previous research in this field, the experimental biosynthesis and characterization of silver nanoparticles mediated using *E. americana* bulb as a reducing and stabilizing agent are not fully understood. Thus, the present work aimed to biosynthesize silver nanoparticles from *E. americana* bulb-mediated extract (Ea-AgNPs), characterize the resulting Ea-AgNPs, evaluate their phytochemical content, and determine the DPPH radical scavenging assay for the Ea-AgNPs.

#### MATERIALS AND METHODS

# Chemicals and extract preparation

All chemicals, such as AgNO<sub>3</sub>, were obtained from Sigma Aldrich, MO, USA. The *E. americana* bulbs were purchased from a local herbal market in Samarinda, East Kalimantan, Indonesia. The identity of the *E. americana* 

was confirmed by a taxonomist working at Universitas Mulawarman. To obtain bulb extract of *E. americana*, the bulb was cleaned using distilled water, thinly chopped into small pieces, and soaked using ethanol (10 g per 100 mL). After 48 hours, the mixture was filtered and vacuum dried using a rotary evaporator. Finally, a dark brownish color *E. americana* bulb ethanolic extract was obtained and stored at 4°C until further use.

#### Ea-AgNPs synthesis

To synthesize the Ea-AgNPs, 10 mL of AgNO<sub>3</sub> solution at various concentrations (0.5, 1, 2, 3, and 4 mM) was prepared and added to 1 mL of *E. americana* bulb ethanolic extract. The synthesis was performed for 24 h at room temperature in a dark bottle to minimize photo-activation of AgNO<sub>3</sub>. The color shift of the solution from colorless to brown confirmed the reduction of Ag to Ag0; UV-Visible spectroscopy was also used to validate its formation.

#### **Ea-AgNPs** characterization

The Ea-AgNPs which was obtained from maximum absorbance (1 mL extract of *E. americana* and 10 mL of AgNO<sub>3</sub> 4mM) was freeze-dried and characterized by using UV-Visible Spectroscopy (UV-VIS; Spectrophotometer SP-UN52N, Beijing China), Scanning Electron Microscopy (SEM; JEOL, Tokyo, Japan), Transmission Electron Microscopy (TEM; JEM 2100F JEOL, Tokyo, Japan), Fourier-Transform Infrared Spectroscopy (FTIR; 1000 FT-IR spectrometer, Perkin Elmer), and X-ray Fluorescence (XRF; EDAX Bruker, Bavaria, Germany).

# Phytochemicals analysis of Ea-AgNPs

The presence of phenols, saponins, triterpenes, flavonoids, alkaloids, and steroids in the Ea-AgNPs was determined using phytochemical screening. The screening process was carried out according to the technique described previously by Dada et al. (2018) and Senguttuvan et al. (2014).

# **Antioxidant activities**

The antioxidant activity of Ea-AgNPs was determined using the DPPH radical scavenging test. The Ea-AgNPs' capacity to scavenge free radicals has previously been evaluated using the stable radical DPPH approach (Lateef et al. 2015). One milliliter of Ea-AgNPs in methanol at different concentrations (0.08, 0.1, 0.12, 0.14, and 0.16 mg/mL) was added to four milliliters of a 0.1 mmol L<sup>-1</sup> methanolic solution of DPPH. A blank was also prepared by diluting 1 mL methanol in 4 mL DPPH. The samples were incubated for 30 minutes in the dark at room temperature. The absorbance (517 nm) was measured against the prepared blank. The percent inhibition of free radicals by DPPH was estimated using the following formula:

Inhibition 
$$\% = \frac{\text{(Acontrol-A sample)}}{\text{Acontrol}} \times 100$$

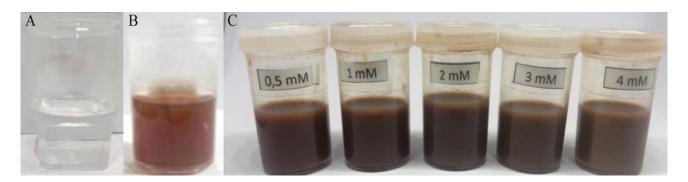
Where  $A_{control}$  is the absorbance of the control reaction (containing all reagents except the test compound) and  $A_{sample}$  is the absorbance of the test compound.

#### RESULTS AND DISCUSSION

### Biosynthesis of Ea-AgNPs

The addition of *E. americana* bulb extract to aqueous AgNO<sub>3</sub>solution resulted in a change in the color of the solution from dark yellowish to reddish-brown (Figure 1) during the reaction, owing to the excitation of surface plasmon vibrations in the silver nanoparticles (Ider et al. 2017). Furthermore, silver nanoparticles were produced at various AgNO<sub>3</sub> concentrations with ethanolic extract of *E. americana* bulb, as evidenced by the plasmon resonance band found at 410 nm within the UV spectra (corresponding to Ea-AgNPs); AgNO<sub>3</sub> had an absorbance peak at 300 nm and the ethanolic extract of *E. americana* bulb had a peak at between 500-600 nm. These findings are

comparable with those of Pandian et al. (2015), stating that Ea-AgNPs shows a characteristic peak absorbance wavelength between 400-500 nm (Figure 2). Previous research has also noted that silver nanoparticle absorb light between 400 and 500 nm due to their surface plasmon resonance (Prathna et al. 2011). The optimum peak of 410 nm for Ea-AgNPs which derived from 4 mM AgNO<sub>3</sub> added with *E. americana* bulb extract indicates the biogenesis of Ag nanoparticles (Figure 3). The current work also shows that Ag nanoparticles are extremely stable in solution, even six days after their synthesis, which significantly confirms the suitability of *E. americana* ethanolic bulb extract for the production of AgNPs.



**Figure 1.** A. A solution of AgNO<sub>3</sub>, B. Dark yellowish of ethanolic extract of *Eleutherine americana* bulb, C. Reddish-brown solution of Ea-AgNPs with different mixture concentrations of AgNO<sub>3</sub> (0.5-4 mM)

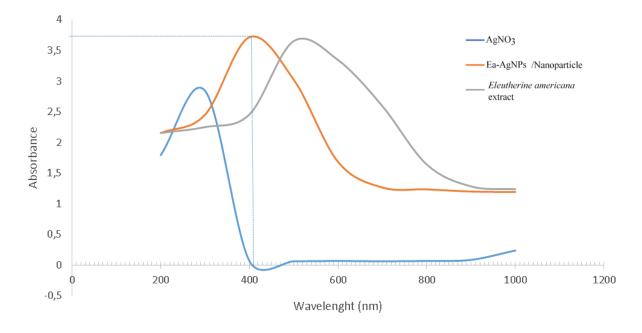


Figure 2. UV-VIS spectra of green synthesis of silver nanoparticles (Ea-AgNPs) using ethanolic extract of *Eleutherine americana* bulb and 4mM of AgNO<sub>3</sub>

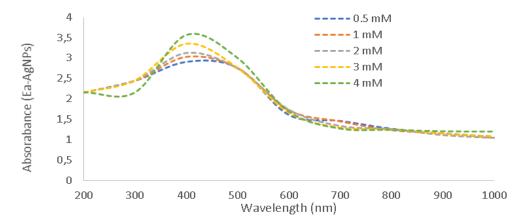


Figure 3. UV/VIS spectra characterization, showing the production of Ea-AgNPs at various concentrations of AgNO<sub>3</sub>

#### **Ea-AgNPs** characterization

The FTIR analysis was used to determine the various functions involved in the reduction of metals to metal nanoparticles. Figure 4 shows the infrared spectrum of biogenic Ea-AgNPs, with four significant bands detected at wavenumbers of 2919, 2850, 1586, and 1031 cm<sup>1</sup>. At 2919 and 2850 cm<sup>1</sup>, a wide peak corresponds to the asymmetric and symmetric stretching v(C-H) vibrations of the methylene group of aliphatic compounds, respectively. The absorption peak at 1586 cm<sup>1</sup>likelycorresponds to the v(-COO) stretching frequency of the carboxyl group, whereas the band at 1031 cm<sup>1</sup> is attributed to the C-O stretching vibration of the carbohydrate residues.

In addition, the X-ray Powder Diffractometry (XRD) pattern indicated that the Ea-AgNPs were amorphous in nature (Figure 5), which is consistent with previously published results (Umadevi et al. 2012). Based on the particle size analysis (Figure 6), the resulting Ea-AgNPs had a size of more than 1 um, which might be caused by agglomeration, and distribution of particles size of Ea-AgNPs were found to be 24.59 % of 6667.10 nm and 75.41 % of 7532.65 nm.

# SEM and EDS Analysis of Ea-AgNPs

The presence of Ea-AgNPs was examined using Energy-Dispersive Spectroscopy (EDS) and SEM (Figure 7). The elemental composition of the Ea-AgNPs' nanocomposite film was determined using Energy Dispersive X-ray analysis (EDX), with the EDX spectrum of the Ea-AgNPs nanocomposite shown in Figure 8. The SEM image of the Ea-AgNPs (Figure 7) demonstrates that almost all of the nanoparticles are widely dispersed throughout the plant extract.

The SEM/EDX was used to determine the nanoparticles' surface morphology and elemental composition (Anake et al. 2016). Figure 7 depicts the morphology of the Ea-AgNPs produced as amorphous aggregates. The EDX analysis was used to determine the elemental components and relative abundance of biosynthesized Ea-AgNPs, as shown in Figure 8. The purity and entire chemical composition of the Ea-AgNPs are shown by the EDX spectrum (Figure 8). The percentage content of Ag metal present in association with

other chemical elements was discovered to be significant. The EDX spectrum revealed an optical absorption characteristic peak corresponding to Ag at 3 keV. Overall, the EDX examination revealed three signals: a strong signal from C atoms (70.99 %), in addition to signals from O atoms (28.95 %) and Ag atoms (0.06 %). Other elements or impurities did not exhibit obvious peaks. Metal silver nanoparticles have a typical optical absorption peak at about 3.7 keV; however, the additional peaks for C and O indicate that the nanoparticles are mixed into precipitates in the plant extract.

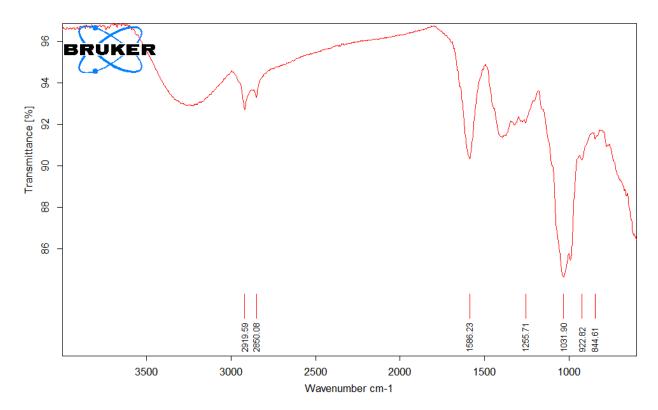
#### **TEM characterization**

Following confirmation of the formation of Ea-AgNPs through color change observation, and UV/VIS absorption spectra, the size, shape, and morphology of Ea-AgNPs were investigated using TEM examination. The TEM images show that the powder particles were amorphous in form and agglomerated (Figure 9). As shown, the Ea-AgNPs were successfully formed, with an average particle size of~1  $\mu m$ .

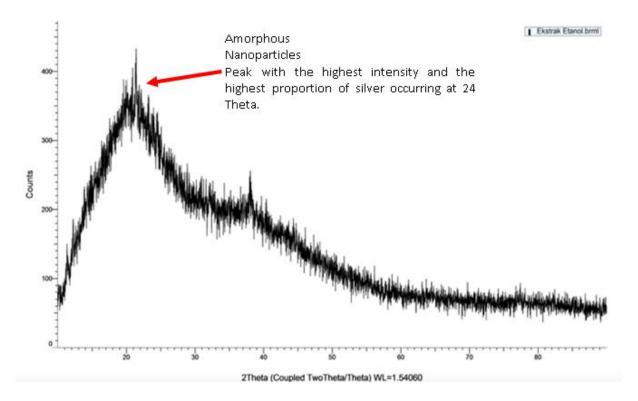
#### The phytochemical contents and DPPH assay

The phytochemical contents in the Ea-AgNPs are summarized in Table 1. The presence of alkaloids, flavonoids, phenolics, tannins, and coumarin was observed in the Ea-AgNPs.

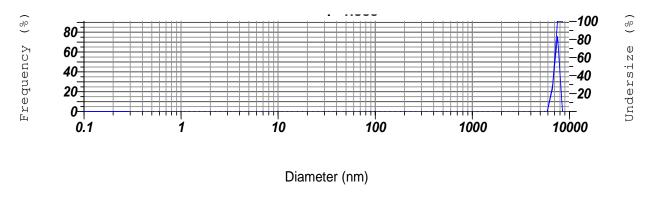
The results indicate that the EA-AgNPs contains several important phytochemicals, including alkaloids, flavonoids, phenolics, tannins, and coumarin (Table 1). These results are similar to findings from previous studies, which revealed that silver nanoparticles from *Sargassum tenerrimum* contain several phytochemicals (Kumar et al. 2012). Some phytochemicals such as flavonoids, tannins, alkaloids, and saponins, which are found in Ea-AgNPs may exhibit antibacterial properties (Rai et al. 2020; Nahar et al. 2021). Meanwhile, DPPH assays were used to assess the antioxidant activity of Ea-AgNPs; the results indicate that the Ea-AgNPs have a higher free radical inhibition percentage, with IC<sub>50</sub> values of 45.30 ppm (Table 2).



**Figure 4.** Fourier-transform infrared spectroscopy (FTIR) spectra of green synthesized silver nanoparticles (Ea-AgNPs) using ethanolic extract of *Eleutherine americana* and 4 mM AgNO<sub>3</sub>; peaks at 2919, 2850, 1586, and 1031 cm<sup>1</sup> were due to C-H methylene group of aliphatic compounds, (-COO) stretching frequency of carboxyl group, and C-O stretching vibration of the carbohydrate residues, respectively



**Figure 5.** The X-ray Powder Diffractometry (XRD) pattern of the green synthesized silver nanoparticles mediated by *Eleutherine americana* (AgNPs); there was no characteristic peak which revealed the amorphous structure of nanoparticles



**Figure 6.** Particle size distribution curve of silver nanoparticles (Ea-AgNPs) complexes which were obtained by particle size analyzer; the Ea-AgNPs was obtained from combination between ethanolic extract of *Eleutherine americana* and 4 mM AgNO<sub>3</sub>

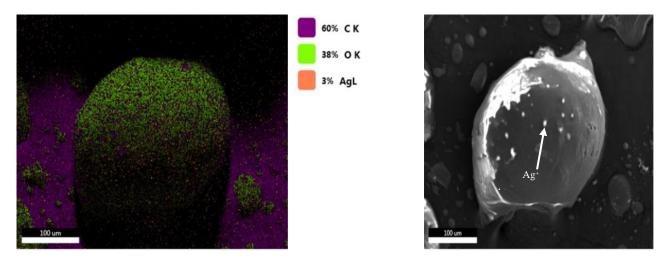


Figure 7. Scanning Electron Microscopy (SEM) images of reduction of Ag<sup>+</sup> to Silver nanoparticles

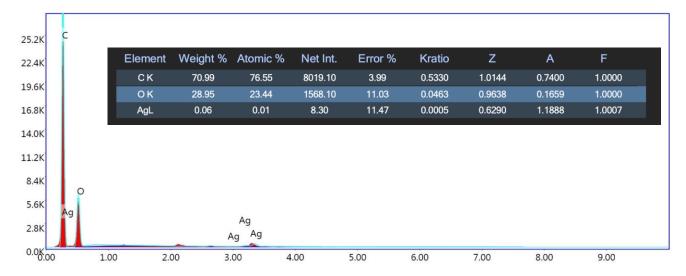
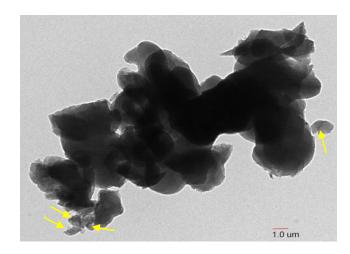


Figure 8. The Energy Dispersive X-ray (EDX) spectrum analysis of Eleutherine americana silver nanoparticle (Ea-AgNPs)



**Figure 9.** Transmission Electron Microscopy (TEM) micrographs of the green synthesized silver nanoparticles (AgNPs), mediated by *Eleutherine americana* bulb aqueous extract

Table 1. Ea-AgNPs phytochemicals content

Phytochemicals	Indicator
Alkaloid	+
Flavonoid	+
Phenolic	+
Saponin	-
Triterpenoid	-
Steroid	-
Tannin	+
Glycoside	-
Coumarin	+
Carotenoid	-

Note: (+) Present, (-) Absent

**Table 2.** Antioxidant activity

Antioxidant (%)	Sample concentration (ppm)
49.67685398	80
57.31063763	100
58.4287385	120
59.37676585	140
64.46053971	160
IC <sub>50</sub>	45.30 ppm

Plants are well-known to contain a considerable number of phenolic and flavonoid compounds, which may have superior anti-oxidative properties and are regarded as potent free radical scavengers (Mustafa et al. 2010). The present  $IC_{50}$  value showed that the Ea-AgNPs have significant antioxidant action. Large molecular weight values and the closeness of multiple aromatic rings and hydroxyl groups are crucial for bioactive compounds' free radical scavenging action (Hagerman et al. 1998). Further, Panneerselvam et al. (2011) demonstrated the in vitro antioxidant capacity of synthesized silver nanoparticles mediated by *Andrographis paniculata* (Acanthaceae) using a DPPH radical scavenging test, which corroborates the findings of our study.

In conclusion, this study demonstrates the successful synthesis of Ea-AgNPs using ethanolic extract of E. americana bulb. An optimum UV/VIS spectrum absorption peak at 400 nm was found, corresponding to the resulting Ea-AgNPs. The FTIR analysis revealed several important biocompounds and XRD characterization found an amorphous Ea-AgNP peak at 24 Theta. Meanwhile, the particle size distribution curve of Ea-AgNPs showed a size of 10<sup>5</sup> nm, with an optical absorption characteristic peak at 3 keV observed via SEM/EDX analysis. The EDX examination revealed three signals: a strong signal from C atoms (70.99 %) and signals from O atoms (28.95 %) and Ag atoms (0.06 %). The TEM imaging also showed the characteristics of Ea-AgNPs. Phytochemical screening tests identified the presence of some phytocompounds such as flavonoids, tannins, alkaloids, and saponins in the Ea-AgNPs and showedIC<sub>50</sub> values of 45.30 ppm.

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