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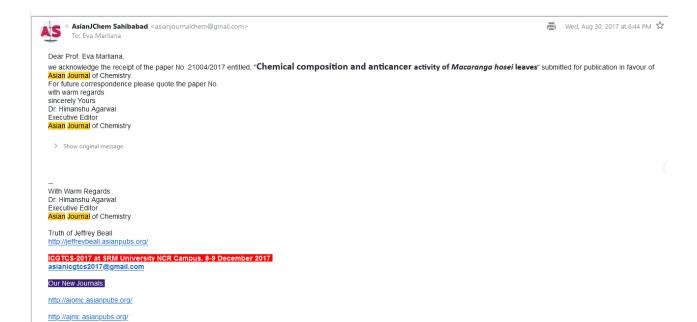


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Dear Dr. Eva Marliana,

We are pleased to inform you that your research article No. 21004/2017 entitled, "Chemical composition and anticancer activity of Macaranga hosei leaves" has been accepted for publication in Asian Journal of Chemistry. The manuscript will appear in Volume 29 (2017) of Asian Journal of Chemistry. Here, we are enclosing an invoice bill No. 1633 dated  $23^{\rm rd}$  November 2017 of US \$ 300=00 (US Dollars Three Hundred Only) towards the Printing/Publication charges.

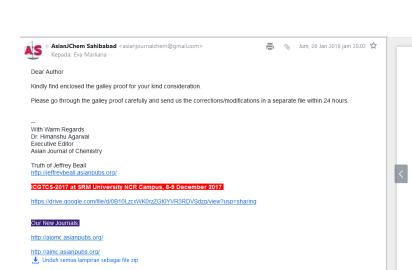
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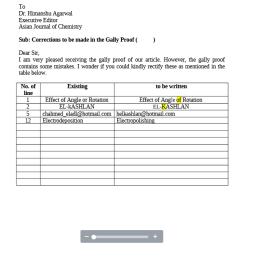


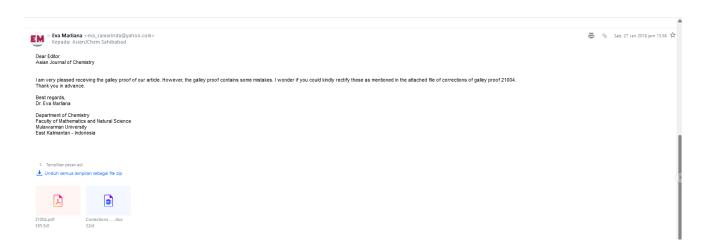
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# Chemical Composition and Anticancer Activity of Macaranga hosei Leaves

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The anticancer activity of MeOH extract and EtOAc fraction of *Macaranga hosei* leaves against HeLa cell lines were evaluated by 3-4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay. Both extracts displayed anticancer activity with IC<sub>50</sub> values of 36.18 and 7.01  $\mu$ M, respectively, which can be suggested that *M. hosei* is a great potential source of anticancer agents. In addition, two isoprenylated flavanones, 4'-*O*-methyl-8-isoprenyleriodictyol (1) and 6-isoprenyleriodictyol (2) have been isolated from EtOAc fraction.

12 The structures of both compounds have been elucidated based on their spectroscopic data, including 1D and 2D NMR spectra.

Keywords: Macaranga hosei, Anticancer, MTT Assay, Flavanones, Isoprenylated.

## INTRODUCTION

Macaranga is one genus of the family Euphorbiaceae comprising of  $\pm$  300 species. In Indonesia, this plant known as "Mahang". The distribution of Macaranga plants is relatively wide, other than Indonesia, also can be found in Africa, Madagascar, Asia, the east coast of Australia and the Pacific islands [1].

According to previous studies, phenolics such as flavonoids and stilbenoids can be isolated from this genus. The uniqueness of flavonoids and stilbenoids from this genus is the presence of terpenoids at aromatic core such as prenyl, geranyl, farnesyl and geranylgeranyl [2,3]. Prenylated flavonoids including flavanone derivatives mostly can be found in *M. triloba*, *M. trichocarpa*, *M. conivera* and *M. lowii* [3-6]. Flavonol derivatives can be obtained from *M. gigantea*, *M. recurvate*, *M. pruinosa*, *M. rizhinoides* and *M. bicolor* [2,5,7-9]. Dihydroflavone derivatives mostly can be attained in *M. conivera*, *M. alnifolia*, *M. pruinosa* and *M. lowii* [6,8,10,11].

Previous studies have revealed that the presence of isoprenoid chains plays an important role for the biological activity of prenylated aromatic compounds which made them possess better bioactivity than their mother compounds without derivatization or modification [12]. An isoprenylated flavanone compound named 4'-O-methyl-8-isoprenyeriodictyol from M. pearsonii displayed an antioxidant activity with IC50 value of 536.89  $\mu$ l [13]. In addition, another isoprenylated flavanones

such as 4'-O-methyl-8-isoprenylnaringenin and lonchocarpol A from *M. hosei* leaves exhibited antioxidant activities with IC<sub>50</sub> values of 1298.0 and 1115.7 μM, respectively [14]. Prenylated flavonoids were reported to have good anticancer effects. Several of these compounds which were isolated from *M. indica*, *M. kurzii* showed cytotoxic activities against cancer cell lines [15,16].

Although numbers of bioactivities have been reported in this genus, the anticancer activity from *M. hosei* leaves extract has not been investigated. In our research, the anticancer activity of MeOH extract and its EtOAc fraction were determined. Moreover, two compounds belong to isoprenylated flavanones named 4'-O-methyl-8-isoprenyleriodictyol (1) and 6-isoprenyleriodictyol (2) have been isolated from the MeOH extract of *M. hosei* leaves.

## **EXPERIMENTAL**

All reagents used were obtained from Merck Chemical, Co. without further additional purification. The isolation were monitored by thin layer chromatography (TLC) and visualized under UV 254 and 356 nm with cerium sulfate as staining agent. Vacuum liquid chromatography (VLC) and radial chromatography were carried out using silica gel 60 GF<sub>254</sub> and silica gel 60 PF<sub>254</sub>. For TLC analysis, pre-coated silica gel plates (Merck Kieselgel 60 GF<sub>254</sub>, 0.25 mm thickness) were used.  $^1\mathrm{H}$  and  $^{13}\mathrm{C}$  NMR spectra were recorded with a JEOL ECS 400 spectro-

meter operating at 400 (<sup>1</sup>H) and 100 (<sup>13</sup>C) MHz in CDCl<sub>3</sub> using TMS as the internal standard. In addition, BioTek PowerWave XSPlate Reader and 5 % CO<sub>2</sub> incubator at 37 °C were also used in this research.

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The leaves of *Macaranga hosei* were collected from Samboja, Kutai Kartanegara, East Kalimantan, Indonesia. This species was identified at the Herbarium of Wanariset, Samboja, Kutai Kartanegara, East Kalimantan, Indonesia and a voucher specimen had been deposited at that herbarium.

**Extraction and isolation:** The dried leaves of *M. hosei* (1.0 kg) were grounded and macerated with MeOH at room temperature and filtered every 2 days. The MeOH crude extract (150 g) was obtained after evaporated by rotary evaporator. Furthermore, the crude was partitioned with n-hexane and EtOAc, respectively. The EtOAc fraction (35 g) was further fractionated by VLC on silica gel with *n*-hexane:EtOAc by increasing the polarity (9:1, 4:1; 7:3, 1:1 and 1:4). Further separation by VLC using n-hexane:EtOAc (9:1 to 3:7), followed by *n*-hexane:CHCl<sub>3</sub> (9:1 to 3:7) using chromatotron yielded compound 1 12.0 mg and compound 2 0.6 mg. Moreover, the structures of these compounds were elucidated by spectroscopic including 1D and 2D NMR.

#### **Determination of anticancer activity**

Cell culture: HeLa cervical cancer cell lines were cultured in the eagle's minimum essential medium containing 1.5 g/L of Na<sub>2</sub>CO<sub>3</sub> and supplemented with 1 % of L-glutamine, 1 % of formulation of antibiotics and antimycotics, 1 % of non-essential amino acids, 1 % of sodium pyruvate and 10 % of fetal bovine serum (FBS). Furthermore, these cells were incubated with 5 % CO<sub>2</sub> incubator at a temperature of 37 °C.

Anticancer activity: The anticancer activities of MeOH extract and EtOAc fraction of M. hosei leaves were determined by method as described by Fahmi et al. [17] with modification, using MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide] assay. HeLa cells were placed in 12 wells containing 25,000 cells/well. After 24 h, cells were washed with phosphate-buffered saline (PBS) and incubated with different concentrations of sample for 24 h. Furthermore, the cells were washed with PBS twice. Then 1 mL of 500 mg/mL MTT was added into each cell and incubated for 4 h. Dark blue formazan crystals formed were then dissolved in  $200\ mL$ of DMSO to measure the absorbance at a ë of 570 nm by BioTek PowerWave XSPlate Reader.

# RESULTS AND DISCUSSION

106 **Anticancer activity:** The anticancer activities of MeOH extract (IC<sub>50</sub> 36.18 µg/mL) and EtOAc extract (7.01 µg/mL) of M. hosei leaves against HeLa cells by MTT assay were found to be active as anticervical cancer. The  $IC_{50}$  values indicated that 109 the anticancer activities of MeOH extract belongs to moderate while EtOAc fraction displayed higher effect than MeOH 111 112 extract. Based on National Cancer Institute, EtOAc fraction signified as an active anticancer due to it has  $IC_{50} \le 30 \,\mu\text{g/mL}$ 114 [18]. The anticancer activity which is showed by M. hosei can 115 be assumed caused by its bioactive compounds, one of them 116 is prenylated flavonoids such as flavanone derivatives which 117 known have a wide variety of biological activities.

**Flavanone derivatives:** Due to EtOAc fraction of *M. hosei* leaves was found to be more active than its MeOH extract, further 119 separation had been conducted to isolate bioactive compounds. 120 Two isolated flavanone derivatives, i.e. 4'-O-methyl-8-isopre- 121 nyleriodictyol (1) and 6-isoprenyleriodictyol (2). The position 122 of protons and carbons of compounds 1 and 2 are presented in 123 Tables 1 and 2, respectively. In addition, HMBC correlations 124 of both compounds are shown in Fig. 1.

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TABLE-1 NMR DATA OF COMPOUND <b>1</b> IN CDCl <sub>3</sub> , 400 MHz					
No. C	$\delta_{\rm H}$ (mult, $J$ in Hz)	$\delta_{\rm C}$	HMBC		
2	5.36 (dd, 12.0, 3.2)	78.5	C-4, C-1', C-2', C-6'		
3	$3.03 (dd, 17.1, 12.0)_{ax}$	42.5	C-2, C-4		
	2.70 (dd, 17.1, 3.2) <sub>eq</sub>		C-1'		
4	-	197.1	-		
4a	-	102.3	-		
5	-	161.6	-		
6	5.98(s)	95.8	C-4a, C-5, C-7, C-8		
7	-	164.8	-		
8	-	107.4	-		
8a	-	160.0	-		
1'	-	132.0	-		
2'	6.87 (d, 2.4)	114.4	C-2, C-4', C6'		
3'	-	146.9	-		
4'	-	148.2	-		
5'	6.89 (d, 8.4)	112.3	C-1', C-3'		
6'	6.82 (dd,8.4, 2.4)	117.8	C-2', C-4'		
1"	3.04 ( <i>d</i> , 7.0)	21.8	C-7, C-8, C-8a, C-2",		
			C-3"		
2"	5.05 (t, 8.6)	123.2	C-3", C-4", C-5"		
3"	-	130.8	-		
4"	1.55 (s)	18.1	C-2", C-3", C-5"		
5"	1.52 (s)	26.0	C-2", C-3", C-4"		
5-OH	12.05 (s)	-	C-4a, C-5, C-6		
7-OH	10.70(s)	-	C-7, C-8		
3'-OH	9.01 (s)	-	C-2', C-4'		
4'-OCH <sub>3</sub>	3.73 (s)	56.1	C-4'		

TABLE-2 NMR DATA FOR COMPOUND <b>2</b> IN CDCl <sub>3</sub> , 400 MHz					
No. C	$\delta_{\rm H}$ (mult, $J$ in Hz)	$\delta_{\rm C}$	HMBC		
2	5.27 (dd, 12.8, 3.2)	78.5	C-2'		
3	$3.02 (dd, 17.2, 12.8)_{ax}$	43.6	C-2, C-4		
	2.79 (dd, 17.2, 3.2) <sub>eq</sub>				
4	-	196.8	-		
4a	-	103.1	-		
5	-	159.6	-		
6	-	107.6	-		
7	-	162.6	-		
8	6.38 (s)	95.8	C-7		
8a	-	161.6	-		
1'	-	129.3	-		
2'	6.98(d, 1.6)	113.7	C-2, C-4', C-6'		
3'	-	147.7	-		
4'	-	144.0	-		
5'	6,89 (d, 8,0)	115.7	C-1', C-3'		
6'	6.87 (dd, 8.0, 1.6)	119.2	C-2, C-4'		
1"	3.30 ( <i>d</i> ,7.2)	21.5	C-5, C-6, C-7, C-2", C-3"		
2"	5.19 (bt)	122.0	C-4", C-5"		
3"	-	135.0	-		
4"	1.81 (s)	18.2	C-2", C-3", C-5"		
5"	1.71 (s)	26.2	C-2", C-3", C-4"		
5-OH	12.32 (s)	-	C-4a, C-5, C-6		

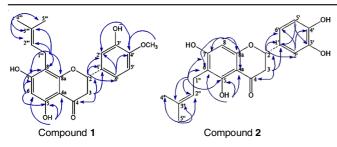


Fig. 1. HMBC correlations of compounds 1 and 2

Compound 1 was obtained as white powder. <sup>1</sup>H NMR 126 127 spectra analysis of compound 1 displayed the characteristic 128 of flavanone such as three protons signals of doublet-doublet at  $\delta_{\rm H}$  5.36 ppm (J = 12.0, 3.2 Hz, H-2), 3.03 ppm (J = 12.0; 129 17.1 Hz, H-3<sub>ax</sub>) and 2.70 ppm (J = 17.1; 3.2 Hz, H-3<sub>eq</sub>). <sup>1</sup>H NMR spectra analysis of compound 4'-O-methyl-8-isoprenyleriodictyol exhibited three aromatic protons signals of ABX system 132 such as doublet signal at  $\delta_{\rm H}$  6.89 ppm (J = 8.4 Hz, H-5'), doublet 133 signal at  $\delta_{\rm H}$  6.87 ppm (J = 2.4 Hz, H-2') and doublet-doublet 134 signal at  $\delta_{\rm H}$  6.82 ppm (J = 8.4, 2.4 Hz, H-6') [3]. This compound 135 136 showed one substituent of isoprenyl (vinyl signal as triplet at  $\delta_{H}$  5.05 ppm; methylene signal as doubletat  $\delta_{H}$  3.04 ppm, two 137 138 methyl signals as singlet at  $\delta_H$  1.55 and 1.52 ppm) and one methoxy signal as singlet ( $\delta_{\rm H}$  3.73 ppm). The presence of one 139 proton singlet signal at  $\delta_H$  5.98 ppm indicated that isoprenyl 140 substituent bonded at C-6 or C-8. 141

Spectra analysis of <sup>13</sup>C NMR from compound 1 showed 21 carbon signals which are distinguished well. The compound consists of six carbons methine, two carbons of methylene, three carbons of methyl and ten quaternary carbons. The carbonyl signal showed at  $\delta_{\rm C}$  197.1 ppm and one signal of oxycarbonmethine was shown at  $\delta_C$  78.5 ppm. Five signals of oxyaryl carbon were shown at  $\delta_C$ : 164.8, 161.6, 160.0, 148.2 and 146.9 ppm. Those signals indicated flavanone with eriodictyol moiety.

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The isoprenyl and methoxy positions of compound 1 were elucidated based on HMQC and HMBC. Long-range correlation between proton signal of 5-OH at  $\delta_{\rm H}$  12.05 ppm with two quarternary carbon atoms ( $\delta_{\rm C}$  161.6 ppm, C-5; 102.3 ppm, C-4a) and one aromatic methine carbon ( $\delta_{\rm C}$  95.8 ppm, C-6) showed that isoprenyl substituent bonded at C-8. Correlation of methoxy proton signal at  $\delta_{\text{H}}$  3.73 ppm with oxyaryl carbon signal ( $\delta_C$  148.2 ppm) displayed that the methoxy group bonded at C-4'.

Based on NMR spectra analysis, it can be elucidated that compound 1 is 4'-O-methyl-8-isoprenyleriodictyol. This compound gave NMR parameterwhich is suitable with 4'-Omethyl-8-isoprenyleriodictyol from M. conifera [5].

Compound 2 was gained as yellow oil. <sup>1</sup>H NMR spectra analysis of compound 2 showed the characteristic of flavanone as well, three protons signals of doublet-doublet at  $\delta_H$  5.27 ppm (J = 12.8, 3.2 Hz, H--2), 3.02 ppm (J = 12.8, 17.2 Hz, H-- $3_{ax}$ ) and 2.79 ppm ( $J = 17.2, 3.2 \text{ Hz}, \text{H-}3_{eq}$ ) and three protons aromatic signals of ABX system at  $\delta_{\rm H}$  6.98 ppm ( $J=1.6~{\rm Hz}$ , H-2'),  $\delta_{\rm H}$  6.89 ppm (J = 8.0 Hz, H-5') and doublet-doublet at  $\delta_{\rm H}$  6.87 ppm (J=8.0, 1.6 Hz, H-6'). The isolated compound displayed one substituent of isoprenyl (vinyl signal as triplet 172 at  $\delta_H$  5.19 ppm, methylene signal as doublet at  $\delta_H$  3.30 ppm and two methyl signals assinglet at  $\delta_H$  1.81 and 1.71 ppm)

together with one aromatic proton signal as singlet in A ring 174 at  $\delta_{\rm H}6.38$  ppm exhibited that isoprenyl bounded at C-6 or C-8. 175

<sup>13</sup>C NMR spectra analysis of compound **2** indicated 20 carbon signals which are separated completely, consist of six methine carbon atoms, two methylene carbon atoms, two methyls and ten quartener carbon atoms. This compound has also eriodictyol structure with one isoprenyl substituent.

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Isoprenyl position of compound 2 was elucidated by HMQC and HMBC. Correlation of long-range between proton signal of 5-OH at  $\delta_H$  12.32 ppm with three quarternary carbon atoms signals ( $\delta_{\rm C}$  159.6 ppm, C-5; 107.6 ppm, C-6;103.1 ppm, C-4a) indicated the presence of isoprenyl substituent at C-6.

Based on data of NMR (including 1D and 2D), compound 2 was elucidated as 6-isoprenyleriodictyol. <sup>1</sup>H and <sup>13</sup>C NMR spectra of the isolated compound is similar with 6-isoprenyleriodictyol which has a molecular structure as C<sub>20</sub>H<sub>20</sub>O<sub>6</sub> and positive ion mass m/z [M]<sup>+</sup> 356.126 [19].

#### Conclusion

Two isoprenylated flavanones named 4'-O-methyl-8-iso- 192 prenyleriodictyol (1) and 6-isoprenyleriodictyol (2) have been 193 isolated from the MeOH extract of M. hosei leaves. In addition, 194 the present study revealed that MeOH extract and EtOAc fraction 195 of M. hosei leaves exhibited significant anticancer activity against HeLa cell lines. It can be suggested that M. hosei is a great potential source as anticancer agents and assumed that two isolated compounds belong to isoprenylated flavanones may play important role in anticancer property.

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#### REFERENCES

- J.W.F. Slik Priyono and Welzen V.P.C., Gard. Bull. (Singapore), 52, 11 1. (2000).
- M. Tanjung, E.H. Hakim and L.J.M. Elfahmi, Nat. Prod. Commun., 7, 1309 (2012).
- Y.M. Syah, E.H. Hakim, S.A. Achmad, M. Hanafi and E.L. Ghisalberti, Nat. Prod. Commun., 4, 63 (2009).
- I. Zakaria, N. Ahmat, F.M. Jaafar and A. Widyawaruyanti, Fitoterapia,
- https://doi.org/10.1016/j.fitote.2012.04.020.
- M.A. Versiani, T. Diyabalanage, R. Ratnayake, C.J. Henrich, S.E. Bates, J.B. McMahon and K.R. Gustafson, J. Nat. Prod., 74, 262 (2011); https://doi.org/10.1021/np100797y
- Agustina, W., Juliawaty, L.D., Hakim, E.H. and Syah, Y.M., ITB J. Sci., 44 A. 13 (2012).
- M. Tanjung, E.H. Hakim, D. Mujahidin, M. Hanafi and Y.M. Syah, J. Asian Nat. Prod. Res., 11, 929 (2009); https://doi.org/10.1080/10286020903302315.
- Y.M. Syah and E.L. Ghisalberti, Nat. Prod. Commun., 5, 219 (2010).
- M. Tanjung, D. Mujahidin and E.H. Hakim, Nat. Prod. Commun., 5, 1209 (2010).
- D.S. Jang, M. Cuendet, A.D. Pawlus, L.B.S. Kardono, K. Kawanishi, N.R. Farnsworth, H.H.S. Fong, J.M. Pezzuto and A.D. Kinghorn, Phytochemistry, 65, 345 (2004); https://doi.org/10.1016/j.phytochem.2003.10.026.
- B.J. Yoder, Dissertation. Faculty of the Virginia Polytechnic Institute and State University, Blacksburg, Virginia (2007).
- B. Botta, G.D. Monache, P. Menendez and A. Boffi, Trends Pharmacol. Sci., 26, 606 (2005): https://doi.org/10.1016/j.tips.2005.09.012.

# 21004

- 13. E. Marliana, T.S. Tjahtjandarie and M. Tanjung, *J. Kimia Mulawarman*, **3**, 97 (2016).
- E. Marliana, T.S. Tjahtjandarie and M. Tanjung, *Der Pharmacia Letter*, 7, 153 (2015).
- D.S. Yang, W.B. Peng, Y.P. Yang, K.C. Liu, X.L. Li and W.L. Xiao, Fitoterapia, 103, 187 (2015); <a href="https://doi.org/10.1016/j.fitote.2015.04.002">https://doi.org/10.1016/j.fitote.2015.04.002</a>.
- D.S. Yang, J.G. Wei, W.B. Peng, S.M. Wang, C. Sun, Y.P. Yang, K.C. Liu and X.L. Li, *Fitoterapia*, 99, 261 (2014); https://doi.org/10.1016/j.fitote.2014.10.003.
- M.Z. Fahmi, K. Ou, J. Chen, M. Ho, S. Tzing and J. Chang, RSC Advances, 4, 32762 (2014); <a href="https://doi.org/10.1039/C4RA05785F">https://doi.org/10.1039/C4RA05785F</a>.
- C. Bézivin, S. Tomasi, F. Lohézic-Le Dévéhat and J. Boustie, *Phytomedicine*, 10, 499 (2003); https://doi.org/10.1078/094471103322331458.
- S.S. Azimova and V.I. Vinogradova, "Physicochemical and Pharmacological Properties of Flavonoids in Natural Compounds-Flavonoids, Springer Science Bussiness Media, New York, 158 (2013).