Cytotoxic Constituents from the Stem Bark of Chisocheton pentandrus

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Abstract – Eight cytotoxic constituents, consisting of six triterpenoids, cabralealactone (1), cabraleadiol (2), prototiamin A (3), 23-desmethyllimocin B (5), melianodiol (7) and indicalilacol (8) along with one limonoid, neemfruitins A (4) and one protolimonoid, protoxylocarpin G (6), were isolated from the extract of *n*-hexane of the stembark of *Chisocheton pentandrus*. The chemical structures were identified on the basis of spectroscopic evidence and compared to previously reported spectra. These isolated compounds appear for the first time in the plant. Compounds 1 - 8 were evaluated for their cytotoxic effect against MCF-7 breast cancer lines *in vitro*. Among the isolated compounds, melianodiol (7) showed the strongest cytotoxic activity with IC₅₀ values of 16.8 μM. **Keywords** – *Chisocheton pentandrus*, MCF-7 cell lines, Meliaceae, limonoid, triterpenoids

Introduction

The genus *Chisocheton* has approximately 150 species that are widely distributed throughout the tropical regions such as Thailand, Malaysia, and Indonesia. Many *Chisocheton* plants have been traditionally used by local people to treat various diseases, and previous studies on the chemical constituents of the this genus have led to the isolation and determination of a number of compounds, such as triterpenoids and limonoids. In addition, biological effects of the constituents have been reported, such as cytotoxic, anti-inflammatory, antifungal, antimalarial and antimycobacterial effects. During the course of our continuing search for anticancer substances from Indonesia *Chisocheton* plants, 1,7,16,17 the methanol extract of the stem bark of *Chisocheton pentandrus* was found to inhibit significant cytotoxic against MCF-7 breast cancer

lines.

C. pentandrus is distributed in Sulawesi and Sumatera islands of Indonesia and previous investigation have led to the isolation of cytotoxic triterpenoids and limonoids.^{17,18} In the further search for cytotoxic constituent from this plant, we investigated the *n*-hexane extract of the bark of *C. pentandrus* and obtained six triterpenoids and two limonoids (Fig. 1). The cytotoxic effects of these compounds against MCF-7 breast cancer cells were evaluated. Here, we described the structural elucidation of the isolates and their cytotoxic activity.

Experimental

General experimental procedures – The IR spectra were obtained in KBr on a SHIMADZU IR Prestige-2. Mass spectra were measured with a Water QTOF-HR-TOFMS-XEV^{otm} mass spectrometer. NMR spectra was recorded using a JEOL ECZ-500 and ECZ-600 spectrometer using tetramethyl silane (TMS) as an internal standard. Furthermore, chromatographic separations was carried out on silica gel 60 (70-230 mesh and 230-400 mesh, Merck). The PTLC glass plates were precoated

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Fig. 1. Chemical structures of compounds 1 - 8.

with silica gel GF_{254} (Merck, 0.25 mm). Similarly, the TLC plates were precoated with silica gel GF_{254} (Merck, 0.25 mm), and the detection was carried out using 10% H_2SO_4 in ethanol, followed by heating.

Plant materials – The stem bark of *C. pentandrus* was collected at Bogor Botanical Garden, West Java Province, Indonesia in June 2016. The plant was identified by the staff of the Herbarium Bogoriense, Bogor and an exhibition specimen (No. Bo-104) was deposited in the herbarium.

Extraction and isolation – The dried stem bark of *C. pentandrus* (1.8 kg) was grounded to powder and extracted with methanol (4 × 4 L, 4 days each) at room temperature, and concentrated using a rotary evaporator, yielding a concentrated extract (340 g). Furthermore, about 300 g of the methanol extract was suspended in water (600 mL) and successively partitioned with *n*-hexane, ethyl acetate and *n*-butanol. The mixture was then evaporated under reduced pressure, resulting in 10.9, 25.2, and 228.6 g of crude extracts, respectively. In addition, 10.0 g of the *n*-hexane fraction was column chromatographed (CC) using silica gel (70-230 mesh), and eluted with *n*-hexane-ethyl acetate-methanol (10% stepwise), resulting to eight fraction groups (A-H).

Fraction B (439 mg) was subjected to silica gel CC (50 g, 230-400 mesh) and eluted with *n*-hexane-dichloro-

methane-ethyl acetate (5% stepwise) to give seven subfraction groups (B1-B7). Subsequently, part of the group, B3 (230 mg) was purified by crystallization in methanol to yield 1 (86.2 mg), while subfraction B7 (132.3 mg) was chromatographed using silica gel CC (10 g, 230-400 mesh) with gradient elution of n-hexane- dichloromethaneethyl acetate (5% stepwise) to produce seven subfractions (B7A-B7F). Therefore, subfraction B7B (13.2 mg) was separated over silica gel CC (5 g, 230-400 mesh), and eluted with *n*-hexane: dichloromethane (1:1) to yield 2 (2.3 mg). Similarly, subfraction B7C (43.2 mg) was separated over silica gel CC (6 g, 230-400 mesh), and eluted with *n*-hexane-dichloromethane-ethyl acetate (5: 4.5:0.5) to produce 3 (14.2 mg), while Fr. B7D (20.1 mg) was purified by silica gel CC (5 g, 230-400 mesh), and eluted with *n*-hexane: dichloromethane (6:4) to generate **4**

Fraction C (700 mg) was separated over silica gel CC (70 g, 70-230 mesh) with a gradient eluent of *n*-hexane-dichloromethane-ethyl acetate (5% stepwise) to produce five subfractions (C1-C5). Therefore, subfraction C3 (300 mg) was further separated over silica gel CC (30 g, 230-400 mesh), and eluted with *n*-hexane: dichloromethane (4:1) to yield four subfractions (C3A-C3D). Subsequently, subfraction C3C (110 mg) was purified by silica gel CC (15 g, 230-400 mesh), and eluted with dichloromethane:

ethyl acetate (7.5:2.5) to generate **5** (14.2 mg), similarly, subfraction C3D (14.5 mg) was separated over silica gel CC (5 g, 230-400 mesh), and eluted with chloroform: ethyl acetate (6:4) for the production of **6** (1.9 mg).

Fraction D (725 mg) was separated on CC silica gel (80 g, 70-230 mesh) using an elution gradient consisting of *n*-hexane-chloroform-ethyl acetate (5% stepwise) to generate seven sub-fractions (D1-D7). Furthermore, sub-fraction D4 (141.5 mg) was separated with CC (15 g, 230-400 mesh), and eluted with *n*-hexane: chloroform: ethyl acetate (7:1.5:1.5), to yield five subfractions (D4A-D4E), while subfraction D4C (41.2 mg) was purified on CC (5 g, 230-400 mesh), and eluted with dichloromethane-chloroform-ethyl acetate (6.0:1.5:2.5) to produce **7** (11.2 mg) and eluted with dichloromethane: ethyl acetate (7.5:2.5) to obtain **8** (2.1 mg).

Cabralealactone (1) – colorless crystals; m.p. 138-140 °C; IR (KBr) v_{max} cm⁻¹: 2937, 2870, 1754, 1720, 1464, 1379, 1280, 1056; ${}^{1}\text{H-NMR}$ (CDCl₃, 500 MHz): δ_{H} 2.52 (1H, d, J = 10 Hz, H-23a), 2.01 (1H, m, H-22b), 1.95 (1H, m, H-5), 1.91 (1H, m, H-12b), 1.90 (1H, m, H-15a), 1.62 (1H, d, J = 10 Hz, H-23b), 1.56 (1H, m, H-7b), 1.53 (1H, m, H-13), 1.52 (1H, m, H-16a), 1.51 (1H, m, H-11b), 1.50 (1H, m, H-1b), 1.49 (1H, m, H-12a), 1.47 (1H, m, H-22a), 1.47 (1H, m, H-2b), 1.41 (1H, dd, J = 2.4, 13.2 Hz, H-9), 1.40 (1H, m, H-2a), 1.39 (1H, m, H-7a), 1.37 (1H, m, H-6b), 1.33 (3H, s, CH₃-21), 1.32 (1H, m, H-6a), 1.23 (1H, m, H-17), 1.20 (1H, m, H-11a), 1.17 (1H, m, H-1a), 1.15 (1H, m, H-16b), 1.10 (1H, m, H-15b), 0.92 (3H, s, CH₃-18), 0.91 (3H, s, CH₃-28), 0.87 (3H, s, CH₃-30), 0.82 (3H, s, CH₃-19), 0.81 (3H, s, CH₃-29); ¹³C-NMR (CDCl₃, 125 MHz), Table 1; HR-TOFMS, m/z 415.3311 [M+H] +, (calcd. $C_{27}H_{42}O_3$ m/z 414.3134).

Cabraleadiol (2) – colorless needle crystals; m.p. 178-180°C; v_{max} cm⁻¹: 3450, 2942, 1471, 1387, 1075; ¹H-NMR (CDCl₃, 500 MHz): $\delta_{\rm H}$ 4.33 (1H, ddd, J = 2.4, 6.8, 9.6 Hz, H-3), 3.62 (1H, dd, J = 4.8, 10.2 Hz, H-24), 2.62 (1H, m, H-23b), 2.01 (1H, m, H-22b), 1.90 (1H, m, H-15b), 1.85 (1H, m, H-23a), 1.83 (1H, m, H-17), 1.75 (1H, m, H-12b), 1.62 (1H, m, H-13), 1.55 (1H, m, H-2b), 1.53 (1H, m, H-11b), 1.51 (1H, m, H-16b), 1.50 (1H, m, H-7b), 1.49 (1H, m, H-12a), 1.46 (1H, m, H-16a), 1.44 (1H, dd, J = 2.4, 13.2 Hz, H-9), 1.42 (1H, m, H-1b), 1.39 (1H, m, H-6b), 1.38 (1H, m, H-2a), 1.34 (1H, m, H-7a), 1.32 (1H, m, H-6a), 1.24 (1H, m, H-5), 1.22 (1H, m, H-22a), 1.20 (1H, m, H-11a), 1.17 (1H, m, H-1a), 1.17 (3H, s, CH₃-26), 1.13 (3H, s, CH₃-21), 1.09 (3H, s, CH₃-27), 1.04 (1H, m, H-15a), 0.95 (3H, s, CH₃-18), 0.92 (3H, s, CH₃-28), 0.87 (3H, s, CH₃-30), 0.84 (3H, s, CH₃-19), 0.82 (3H, s, CH₃-29); ¹³C-NMR (CDCl₃, 125 MHz), Table 1; HR-

TOFMS m/z 461.3271 [M+H] +, (calcd. $C_{30}H_{52}O_3$ m/z 460.3916).

Prototiamin A (3) - White amorphous powder; IR (KBr) v_{max} cm⁻¹: 3430, 2945, 1720, 1639, 1456, 1387, 1074, 847; 1 H-NMR (CDCl₃, 600 MHz): δ_{H} 7.14 (1H, d, J = 10.2 Hz, H-1), 5.86 (1H, d, J = 10.2 Hz, H-2), 5.28 (1H, br.s, H-15a), 5.05 (1H, m, H-7), 4.47 (1H, dd, J = 9.6, 1.8 Hz, H-23), 3.92 (1H, d, J = 9.6 Hz, H-24), 2.71 (1H, m, H-20), 2.53 (1H, m, H-6b), 2.25 (1H, m, H-6a), 2.20 (1H, m, H-5), 2.19 (1H, m, H-17), 2.19 (1H, m, H-9), 2.05 (1H, m, H-22b), 1.94 (3H, s, CH₃-2'), 1.82 (1H, m, H-22a), 1.82 (1H, m, H-16b), 1.76 (1H, m, H-12b), 1.63 (1H, m, H-13), 1.57 (1H, m, H-12a), 1.50 (1H, m, H-11b), 1.44 (3H, s, CH₃-27), 1.35 (3H, s, CH₃-26), 1.22 (1H, m, H-11a), 1.17 (1H, m, H-16a), 1.16 (3H, s, CH₃-30), 1.16 (3H, s, CH₃-18), 1.07 (3H, s, CH₃-19), 1.06 (3H, s, CH₃-28), 1.06 (3H, s, CH₃-29); ¹³C-NMR (CDCl₃, 150 MHz), Table 1. HR-TOFMS, *m/z* 543.1260 [M+H]⁺, (calcd. $C_{32}H_{46}O_7$ m/z 542.1244).

Neemfruitin A (4) – Colorless solid; IR (KBr) v_{max} cm⁻¹: 3360, 2939, 1725, 1710, 1458, 1380, 1109; ¹H-NMR (CDCl₃, 600 MHz): $\delta_{\rm H}$ 5.34 (1H, t, J = 2.4 Hz, H-1), 5.33 (1H, t, J = 2.4 Hz, H-7), 5.28 (1H, dd, J = 1.8, 3.6 Hz, H-15), 4.57 (1H, dd, J = 9.6, 1.8 Hz, H-23), 4.04 (2H, d, J = 9.6 Hz, H-21), 2.70 (1H, m, H-20), 2.53 (1H, m, H-6b), 2.50 (1H, dd, J = 8.0, 10.4 Hz, H-5), 2.25 (1H, m, H-6a), 2.20 (1H, m, H-9), 2.17 (1H, m, H-22a), 2.16 (1H, m, H-22b), 2.04 (1H, m, H-16a), 1.97 (3H, s, CH₃-2"), 1.93 (1H, m, H-2a), 1.93 (1H, m, H-11a), 1.91 (3H, s, CH₃-2'), 1.73 (1H, m, H-12b), 1.70 (1H, dt, J = 3.0, 10.2 Hz, H-17), 1.61 (1H, m, H-2b), 1.56 (1H, m, H-11b), 1.54 (1H, m, H-12a), 1.45 (1H, m, H-16b), 1.16 (3H, s, CH₃-30), 1.16 (3H, s, CH₃-19), 1.09 (3H, s, CH₃-29), 1.06 (3H, s, CH₃-28), 1.00 (3H, s, CH₃-18); ¹³C-NMR (CDCl₃, 150 MHz), Table 1. HR-TOFMS, m/z 517.4498 [M+H]⁺, (calcd. C₃₀H₄₄O₇ m/z 516.4087).

23-Desmethyllimocin B (**5**) – white solid; IR (KBr) v_{max} cm⁻¹: 3300, 2949, 1705, 1620, 1457, 1380, 1080; ${}^{1}\text{H-NMR}$ (CDCl₃, 600 MHz): δ_{H} 7.14 (1H, d, J = 10.2 Hz, H-1), 5.86 (1H, d, J = 10.2 Hz, H-2), 5.33 (1H, t, J = 2.4 Hz, H-7), 5.28 (1H, dd, J = 1.8, 3.6 Hz, H-15), 4.57 (1H, dd, J = 9.6, 1.8 Hz, H-23), 4.04 (2H, d, J = 9.6 Hz, H-21), 2.70 (1H, m, H-20), 2.53 (1H, m, H-6b), 2.50 (1H, dd, J = 8.0, 12.4 Hz, H-5), 2.25 (1H, m, H-6a), 2.20 (1H, m, H-10), 2.17 (1H, m, H-22a), 2.16 (3H, s, CH₃-2'), 2.16 (1H, m, H-22b), 2.04 (1H, m, H-16a), 1.93 (1H, m, H-11a), 1.92 (1H, m, H-16b), 1.73 (1H, m, H-12b), 1.70 (1H, dt, J = 3.0, 10.2 Hz, H-17), 1.58 (1H, m, H-11b), 1.54 (1H, m, H-12a), 1.16 (3H, s, CH₃-19), 1.06 (3H, s, CH₃-29), 1.06 (3H, s, CH₃-30), 1.00 (3H, s, CH₃-18); ${}^{13}\text{C}$ -

Table 1. ¹³C-NMR data for compounds 1 - 8

Position carbon	Compounds							
	1 * δ _C (mult.)	2^* $\delta_{\rm C}$ (mult.)	$3*$ $\delta_{\rm C}$ (mult.)	$4*$ $\delta_{\rm C}$ (mult.)	$5**$ $\delta_{\rm C}$ (mult.)	$6**$ $\delta_{\rm C}$ (mult.)	7^{**} $\delta_{\rm C}$ (mult.)	$8**$ $\delta_{\rm C}$ (mult.)
1	35.2 (t)	33.7 (t)	158.8 (d)	75.4 (d)	158.8 (d)	158.8 (d)	31.2 (t)	31.2 (t)
2	33.7 (t)	25.4 (t)	125.7 (d)	25.6 (t)	125.7 (d)	125.7 (d)	25.6 (t)	25.6 (t)
3	218.2 (s)	73.2 (d)	204.6 (s)	218.3 (s)	204.6 (s)	204.6 (s)	218.4 (s)	72.4 (d)
4	37.3 (s)	37.3 (s)	44.6 (s)	39.9 (s)	44.6 (s)	44.6 (s)	40.3 (s)	40.3 (s)
5	55.4 (d)	49.6 (d)	46.2 (d)	49.8 (d)	49.8 (d)	46.2 (d)	44.6 (d)	44.6 (d)
6	18.3 (t)	18.3 (t)	27.5 (t)	27.5 (t)	27.5 (t)	27.5 (t)	24.8 (t)	24.8 (t)
7	34.7 (t)	35.4 (t)	75.5 (d)	75.6 (d)	74.4 (d)	34.9 (t)	118.8 (d)	118.8 (d)
8	40.6 (s)	40.7 (s)	42.8 (s)	43.0 (s)	43.0 (s)	42.8 (s)	145.3 (s)	145.3 (s)
9	50.4 (d)	50.7 (d)	37.4 (d)	37.4 (d)	37.4 (d)	37.4 (d)	48.9 (d)	48.9 (d)
10	37.7 (s)	37.7 (s)	39.7 (s)	47.2 (s)	47.2 (s)	39.7 (s)	34.6 (s)	34.6 (s)
11	25.4 (t)	21.7 (t)	34.1 (t)	16.2 (t)	16.2 (t)	34.1 (t)	17.4 (t)	17.4 (t
12	21.3 (t)	27.1 (t)	38.3 (t)	40.0 (t)	40.0 (t)	38.3 (t)	31.2 (t)	31.2 (t
13	43.2 (d)	42.8 (d)	46.8 (s)	50.2 (s)	50.2 (s)	46.8 (s)	43.8 (s)	43.8 (s
14	50.3 (s)	50.2 (s)	157.9 (s)	158.9 (s)	158.9 (s)	157.9 (s)	50.7 (s)	50.7 (s
15	31.2 (t)	31.5 (t)	119.0 (d)	118.0 (d)	118.0 (d)	119.0 (d)	33.9 (t)	33.9 (t
16	25.1 (t)	25.9 (t)	33.8 (t)	34.6 (t)	34.6 (t)	33.8 (t)	23.9 (t)	23.9 (t
17	49.5 (d)	49.8 (d)	58.2 (d)	55.3 (d)	55.3 (d)	58.2 (d)	47.2 (d)	47.2 (d
18	15.6 (q)	16.2 (q)	16.2 (q)	19.7 (q)	19.7 (q)	16.2 (q)	23.5 (q)	23.5 (q
19	16.1 (q)	16.6 (q)	19.1 (q)	15.3 (q)	15.3 (q)	19.1 (q)	13.1 (q)	13.1 (q
20	90.3 (s)	86.7 (s)	44.7 (s)	37.6 (d)	37.6 (d)	44.7 (s)	40.7 (d)	40.7 (d
21	25.4 (q)	27.3 (q)	178.6 (s)	70.3 (d)	72.5 (d)	72.5 (d)	99.3 (d)	178.3 (s
22	31.3 (t)	35.3 (t)	33.1 (t)	34.2 (t)	34.2 (t)	33.1 (t)	29.6 (t)	29.6 (t
23	29.3 (t)	26.4 (t)	78.6 (d)	99.8 (d)	99.2 (d)	78.2 (d)	86.4 (d)	86.4 (d
24	176.9 (s)	86.3 (d)	72.4 (d)	-	_	74.4 (d)	71.3 (d)	71.3 (d
25	-	70.3 (s)	70.5 (s)	-	-	72.5 (s)	70.3 (s)	70.3 (s
26	-	27.9 (q)	26.9 (q)	-	-	26.9 (q)	26.5 (q)	26.5 (q
27	-	24.1 (q)	23.8 (q)	-	-	23.8 (q)	27.0 (q)	27.0 (q
28	28.4 (q)	28.4 (q)	20.0 (q)	21.4 (q)	-	20.0 (q)	21.6 (q)	21.6 (q
29	22.2 (q)	22.2 (q)	21.4 (q)	27.1 (q)	21.4 (q)	21.4 (q)	27.4 (q)	27.4 (q
30	16.4 (q)	15.6 (q)	27.1 (q)	27.5 (q)	27.1 (q)	27.1 (q)	27.8 (q)	27.8 (q
1′	-	-	170.1 (s)	170.1 (s)	27.5 (q)	170.1 (s)	. 2	
2'	-	-	21.2 (q)	21.1 (q)	170.1 (s)	21.2 (q)		
1''	-	-	. 2	170.0 (s)	. ,	. 2		
2"	-	_		21.0 (q)				

^{*}Measured in 500 MHz for ¹H and 125 MHz for ¹³C

NMR (CD₃OD, 150 MHz), Table 1. HR-TOFMS, m/z 455.0712 [M+H]⁺, (calcd. C₂₈H₄₀O₅ m/z 456.2876).

Protoxylocarpin G (6) – White solid; IR (KBr) v_{max} cm⁻¹: 3420, 2949, 1715, 1645, 1457, 1380, 1080; ¹H-NMR (CD₃OD, 600 MHz): $δ_H$ 7.14 (1H, d, J = 10.2 Hz, H-1), 5.86 (1H, d, J = 10.2 Hz, H-2), 5.28 (1H, br.s, H-15), 4.16 (2H, d, J = 9.6 Hz, H-23), 3.92 (1H, d, J = 9.6 Hz, H-24), 2.71 (H, m, H-20), 2.53 (1H, m, H-6b), 2.25

(1H, m, H-6a), 2.20 (1H, m, H-5), 2.19 (1H, m, H-17), 2.19 (1H, m, H-9), 1.94 (3H, s, CH₃-2'), 1.87 (1H, m, H-16a), 1.78 (1H, m, H-12a), 1.69 (1H, m, H-7b), 1.65 (m, H-7a), 1.57 (1H, m, H-12b), 1.52 (1H, m, H-11a), 1.50 (1H, m, H-11b), 1.44 (3H, s, CH₃-27),1.35 (3H, s, CH₃-26), 1.17 (1H, m, H-16b), 1.16 (3H, s, CH₃-30), 1.16 (3H, s, CH₃-18), 1.01 (3H, s, CH₃-19), 1.06 (3H, s, CH₃-28), 1.06 (3H, s, CH₃-29); ¹³C-NMR (CD₃OD, 150 MHz)

^{**} Measured in 600 MHz for ¹H and 150 MHz for ¹³C

Table 1. HR-TOFMS, m/z 527.2984 [M+H] $^+$, (calcd. $C_{32}H_{48}O_6$ m/z 528.2951).

Melianodiol (7) – Colorless needle crystal; m.p. 216-218°C; IR (KBr) v_{max} cm⁻¹ : 3350, 2935, 1725, 1645, 1457, 1380, 1055; ${}^{1}\text{H-NMR}$ (CDCl₃, 600 MHz): δ_{H} 5.26 (1H, dd, J = 2.6, 5.8 Hz, H-7), 4.63 (1H, d, J = 5.6 Hz, H-21), 4.05 (1H, br.s, H-24), 3.64 (1H, ddd, J = 1.3, 6.2, 8.7 Hz, H-23), 2.67 (1H, ddd, J = 5.7, 8.4, 11.7 Hz, H-20), 2.38 (1H, m, H-22a), 2.34 (1H, m, H-17), 2.24 (1H, m, H-9), 2.20 (1H, ddd, J = 6.2, 8.4, 11.7 Hz, H-22b), 2.09 (1H, d, J = 14.7, H-6a), 1.93 (1H, m, H-2a), 1.78 (1H, m, H-5), 1.76 (1H, m, H-12a), 1.73 (1H, m, H-16b), 1.63 (1H, m, H-11a), 1.61 (1H, m, H-2b), 1.61 (1H, m, H-12b), 1.57 (1H, m, H-16a), 1.53 (1H, m, H-11b), 1.52 (1H, m, H-15b), 1.49 (1H, m, H-1a), 1.41 (1H, m, H-6b), 1.39 (3H, s, CH₃-27), 1.38 (1H, dt, J = 3.2, 12.2 Hz, H-1b), 1.32 (3H, s, CH₃-26), 1.06 (1H, m, H-15a), 1.01 (3H, s, CH₃-30), 0.93 (3H, s, CH₃-18), 0.93 (3H, s, CH₃-28), 0.92 (3H, s, CH₃-29), 0.78 (3H, s, CH₃-19); ¹³C-NMR (CDCl₃, 150 MHz) Table 1. HR-TOFMS, m/z 487.4535 [M+H] $^{+}$, (calcd. $C_{32}H_{50}O_7$ m/z 488.4502).

Indicalization B (8) – White solid; IR (KBr) v_{max} cm⁻¹: 3430, 2930, 1760, 1655, 1420, 1378, 1210, 1110; ¹H-NMR (CDCl₃, 600 MHz): δ_H 5.26 (1H, dd, J = 2.6, 5.8 Hz, H-7), 4.63 (1H, d, J = 5.6 Hz, H-21), 4.32 (1H, d, J = 5.4 Hz, H-3), 4.05 (1H, br.s, H-24), 3.64 (1H, ddd, J = 1.3, 6.2, 8.7 Hz, H-23), 2.67 (1H, ddd, J = 5.7, 8.4, 11.7 Hz, H-20), 2.38 (1H, m, H-22a), 2.34 (1H, m, H-17), 2.24 (1H, m, H-9), 2.20 (1H, ddd, J = 6.2, 8.4, 11.7 Hz, H-22b), 2.09 (1H, d, J = 14.7, H-6a), 1.93 (1H, m, H-2a), 1.78 (1H, m, H-5), 1.76 (1H, m, H-12a), 1.73 (1H, m, H-16b), 1.63 (1H, m, H-11a), 1.61 (1H, m, H-12b), 1.61 (1H, m, H-2b), 1.57 (1H, m, H-16a), 1.53 (1H, m, H-11b), 1.52 (1H, m, H-15b), 1.49 (1H, m, H-1a), 1.41 (1H, m, H-6b), 1.39 (3H, s, CH₃-27), 1.38 (1H, dt, J = 3.2, 12.2 Hz, H-1b), 1.32 (3H, s, CH₃-26), 1.06 (1H, m, H-15a), 1.01 (3H, s, CH₃-30), 0.93 (3H, s, CH₃-28), 0.92 (3H, s, CH₃-29), 0.84 (3H, s, CH₃-18), 0.78 (3H, s, CH₃-19); ¹³C-NMR (CDCl₃, 160 MHz) see Table 1. HR-TOFMS, m/z 487.4535 [M+H]⁺, (calcd. C₃₂H₅₀O₇ m/z 488.4502).

Cytotoxicity assay – The MCF-7 Cells were plated at a density of 5×10^3 cells/well in 96-well plate, incubated at 37° C overnight and treated in triplicate with decreasing concentrations isolated compound from *C. pentandrus* extract. Furthermore, all the extracted were dissolved in DMSO and *Cis*-platin (Sigma) was used as a positive control. After 48 hours of treatment, cell viability was evaluated using the MTT reagent 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide was added according to the manufactured protocols, followed by a 4-hour

incubation. The results were expressed as relative viable cells compared to the controls (untreated cells). The 50% inhibitory concentration (IC $_{50}$) values of antiproliferative activity were calculated and converted into the IC $_{50}$ Dose.

Results and Discussion

The methanol extract of the of *C. pentandrus* stem bark was partitioned successfully with *n*-hexane, ethyl acetate and *n*-butanol. Column chromatography was repeated using silica gel of the soluble fraction in *n*-hexane led to the isolation of eight isolated compounds (Fig. 1). The structures of the isolated compounds were identified by spectroscopic methods including 1D, 2D NMR and HR-TOFMS. To the best our knowledge, compounds 1 - 8, were isolated from *C. pentandrus* for the first time.

Compound 1 was obtained as colorless crystals (MeOH). The molecular composition was determined was C₂₇H₄₂O₃ based on HR-TOFMS ($[M+H]^+$ m/z 415.3311, calcd. for $C_{27}H_{42}O_3$ m/z 414.3334) and NMR analysis (Table 1), and seven degrees of unsaturation is required. Furthermore, IR spectra showed absorption peaks at 2937 and 2870 cm⁻¹ (C-H sp³), 1754 cm⁻¹ (C=O), 1720 (C=O, lactone) 1464 cm⁻¹ (C=C), 1379-1280 cm⁻¹ (gem-dimethyl), and 1056 cm⁻¹ (C-O groups). The ¹H-NMR spectrum showed the presence of six tertiary methyl groups, resonating at δ_H 1.33 (CH₃-21), 0.92 (CH₃-18), 0.91 (CH₃-28), 0.87 (CH₃-30), 0.82 (CH₃-19) and 0.81 (CH₃-29). The aliphatic protons in the upfield region were observed also in the ¹H-NMR spectra. The ¹³C-NMR, HMQC and DEPT 135° spectra of 1 showed the presence of six methyl groups, exhibiting the characteristics of dammarane-type triterpenoid, one oxygenated quaternary carbon at $\delta_{\rm C}$ 90.3 (C-20), one carbonyl lactone at δ_C 176.9 (C-24) and one carbonyl ketone at δ_C 218.2 (C-3). These functions represented two of the total of seven degrees of unsaturation, and the remaining five degrees of unsaturation matched the dammarane-type triterpenoid framework with an addition of a five-membered lactone ring. 19,20 The ¹H-¹H correlations (Fig. 2) from H-1/H-2, H-5/H-6/H-7, H-9/H-11/H-12/H-13, H-15/H-16/H-17 and H-22/H-23 were aided by the presence of the dammarane-type in compound 1. The HMBC cross peaks (Fig. 2) from H-1 $(\delta_{\rm H}\ 1.17\ \text{and}\ 1.50)$ and H-2 $(\delta_{\rm H}\ 1.40\ \text{and}\ 1.47)$ to the carbonyl ketone at C-3 ($\delta_{\rm C}$ 218.2) indicated the presence of a ketone group at C-3. Protons signal H-17 ($\delta_{\rm H}$ 1.23) and CH₃-21 (δ_H 1.33) were correlated to C-20 (δ_C 90.3), whereas methylene protons at C-23 (δ_H 2.52 and 2.62) was correlated to C-24 ($\delta_{\rm C}$ 176.9) indicated the presence of five-membered lactone ring at C-17. The relative

Fig. 2. Selected ¹H-¹H COSY and HMBC correlations of compounds 1, 2, 3 and 5.

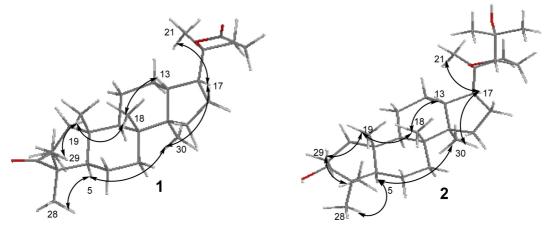


Fig. 3. Selected NOESY correlations for compounds 1 and 2.

configuration of **1** was determined by NOESY experiments, which is supported by the presence of dammarane-type triterpenoids in *Chisocheton* species. ^{19,20} The NOESY correlations (Fig. 3) from H-17/CH₃-30 identified the γ -lactone ring at C-17 as β -oriented, while the cross peak observed between CH₃-30/H-17/CH₃-21, indicates the CH₃-21 as α -oriented. A comparison of the NMR data of **1** with cabralealactone obtained from the whole plant of *Cleome africana*, ¹⁹ revealed that the structures of the two compounds were very similar, consequently, compound **1** was identified as cabralealactone.

Compound **2** was obtained as a colorless needle crystal. Its molecular composition $C_{30}H_{52}O_3$, was determined from HR-TOFMS spectra (m/z 461.3271 [M+H] $^+$, calculated, $C_{30}H_{52}O_3$ m/z 460.3274) and NMR data (Table 1). The IR spectra showed absorption peaks at 3450 cm⁻¹ (OH), 2942 cm⁻¹ (C-H sp³), 1471 and 1387 cm⁻¹ (gem-dimethyl), and 1075 cm⁻¹ (C-O) groups. The 1 H-NMR spectrum showed the presence of eight tertiary methyl groups at δ_H 1.17 (CH₃-26), 1.13 (CH₃-21), 1.09 (CH₃-27), 0.95 (CH₃-18), 0.92 (CH₃-28), 0.87 (CH₃-30), 0.84 (CH₃-19) and 0.82 (CH₃-29) resonances. Two oxygenated

methines resonances at δ_H 4.33 (1H, ddd, J = 2.4, 6.8, 9.6 Hz, H-3) and $\delta_{\rm H}$ 3.62 (1H, dd, J = 4.8, 10.2 Hz, H-24) were also observed in the ¹H-NMR spectra. The ¹³C-NMR spectra showed 30 carbons and were classified by DEPT 135° and HMQC experiments as eight methyl groups, exhibiting the characteristics of dammarane-type triterpenoid, ^{19,20} an oxymethine group at $\delta_C \delta_C$ 86.3 (C-24) and 73.2 (C-3) as well as an oxygenated quaternary carbon at δ_C 70.3 (C-25). The ¹H-¹H correlations (Fig. 2) from H-1/H-2/H-3, H-5/H-6/H-7, H-9/H-11/H-12/H-13, H-15/H-16/H-17 and H-22/H-23 were aided by the presence of the dammarane-type in compound $2^{19,20}$. The HMBC cross peaks (Fig. 2) from CH₃-28 ($\delta_{\rm H}$ 0.92), CH₃-29 ($\delta_{\rm H}$ 0.82), and the methylene protons at H-2 ($\delta_{\rm H}$ 1.55 and 1.38) to the oxymethine carbon at C-3 (δ_C 73.2) indicated the presence of a hydroxy group at C-3. The correlation of H-22 (δ_H 1.22 and 2.01) and H-23 (δ_H 2.52 and 2.62) to C-24 (δ_C 86.3) and C-20 (δ_C 86.7) indicates that the position of furan ring at C-20/C-24. Furthermore, methyl protons at δ_H 1.17 (CH₃-26) and 1.09 (CH₃-27) as well as the oxygenated proton at δ_H 3.62 correlated with oxygenated quaternary carbon at $\delta_{\rm C}$ 86.3 (C-24), indicating that an isopropyl alcohol moiety was located at C-24. In addition, the key NOESY correlation (Fig. 3) between CH₃-21 and CH₃-26, CH₃-27 specifies the isopropyl group at C-24 as a α -orientation, which is supported by the presence of dammarane-type triterpenoids in Chisocheton species. 19,20 The NMR spectra of 2 were in good agreement with that reported cabraleadiol isolated from Chisocheton penduliflorus, 20 therefore, compound 2 was identified as a cabraleadiol.

Compound 3 was isolated as a white amorphous powder. Its molecular composition was assigned as $C_{32}H_{46}O_7$ on the basis of the HRTOFMS (m/z 543.1260 $[M+H]^{+}$, (calcd. $C_{32}H_{46}O_7$ m/z 542.1244 NMR data (Table 1), therefore required ten double-bond equivalents. IR spectra showed absorption peaks at 3430 cm⁻¹ (OH), 2945 cm⁻¹ (C-H sp³), 1720 (C=O), 1639 cm⁻¹ (C=C), 1456 and 1378 cm⁻¹ (gem-dimethyl) and 1076 cm⁻¹ (C-O) groups. The ¹H-NMR spectrum showed the presence of seven tertiary methyl groups, resonance at $\delta_{\rm H}$ 1.44 (CH₃-27), 1.35 (CH₃-26), 1.16 (CH₃-30), 1.16 (CH₃-18), 1.07 (CH₃-19), 1.06 (CH₃-28) and 1.06 (CH₃-29). The signal of one acetyl at δ_H 1.94 (CH₃-2'), two oxygenated methines at δ_H 4.47 (1H, dd, J = 9.6, 1.8 Hz, H-23), 3.92 (1H, d, J=9.6 Hz, H-24), three methine sp² at $\delta_{\rm H}$ 7.14 (1H, d, J = 10.2 Hz, H-1), 5.86 (1H, d, J = 10.2 Hz, H-2)and 5.28 (1H, br.s, H-15) were observed also in the ¹H-NMR spectrum. The ¹³C-NMR spectrum (Table 1) showed the presence 32 carbons, which were assigned to

one ketone (δ_C 204.6), one ester (δ_C 170.1), one lactone $(\delta_{\rm C}\ 178.6)$, four sp² carbons ($\delta_{\rm C}\ 158.8$, 157.9, 125.7 and 119.0), eight methyls (δ_C 27.1, 26.9, 23.8, 21.4, 21.2, 20.0, 19.1 and 16.2), six methylenes and seven methines. These properties represented a total of five or ten degrees of unsaturation, and the remaining five degrees of unsaturation corresponded to the pentacyclic framework of the apotirucallan-type triterpenoid.²¹ The proton pairing was also confirmed with the ¹H-¹H COSY spectrum (Fig. 2) and supported the presence of apotirucallan-type triterpenoid in 3. The HMBC crosspeaks (Fig. 2) from H-2 (δ_H 5.86) to C-1 (δ_C 158.8) and carbonyl at C-3 (δ_C 204.6), indicated the presence of α,β -unsaturated ketone at C-1, C-2 and C-3, respectively. Correlation which was arising from H-7 (δ_H 5.33) to C-1' (δ_C 170.1) and C-6 (δ_C 27.5) as well as from CH₃-2" (δ_H 1.94) to to C-1' (δ_C 170.1), indicate that acetyl group was attached at C-7. The correlation from H-15 (δ_H 5.28) to C-16 (δ_C 33.8) and C-14 ($\delta_{\rm C}$ 157.9), indicated that another double-bond was located at C-14 and C-15, respectively. An oxygenated proton at H-23 (δ_H 4.47) and methine proton at H-17 (δ_H 2.19) was correlated to carbonyl at C-21 ($\delta_{\rm C}$ 178.6), indicate the lactone ring was attached at C-17. Furthermore, the correlation from CH₃-26 (δ_H 1.35) to C-24 (δ_C 72.4) and C-25 ($\delta_{\rm C}$ 70.5), suggesting that 1,2-diol side chain was attached at C-23. The detailed examination of the NMR data and comparison with those reported for prototiamin A, an apotirucallane-type triterpenoid isolated from the bark of Entandrophragma congoense, 21 showed that the structures of the two compounds were very similar. Therefore, compound 3 was identified as prototiamin A.

Compound 4 was isolated as colorless solid, with the molecular formula determined as C₃₀H₄₄O₇ based on HRTOFMS (m/z 517.4498 [M+H]⁺, (calcd. C₃₀H₄₄O₇ m/z516.4087) and NMR spectra (Table 1). In addition, ¹³C NMR analysis show that nine degrees of unsaturation is required, while, the IR absorption bands at 3560, 2939, 1725 and 1109 cm⁻¹ imply the presence of hydroxyl, aliphatic, carbonyl ester and ether functionalities. The ¹H-NMR spectrum showed the presence of five methyl tertiary signals at $\delta_{\rm H}$ 1.16 (CH₃-19), 1.16 (CH₃-30), 1.06 (CH₃-28), 1.06 (CH₃-29), 1.00 (CH₃-18) and two acetyl signals at δ_H 1.91 and 1.97. One olefinic proton at δ_H 5.28 (1H, dd, J=1.8, 3.6 Hz, H-15) and three oxygenated methane signals at $\delta_{\rm H}$ [5.39 (1H, m, H-1), 5.33 (1H, m, H-7) and 4.57 (1H, dd, J=9.6, 1.8 Hz, H-23)] were also observed in the ¹H-NMR spectrum. Moreover, the ¹³C NMR spectrum (Table 1) recognized 30 nonequivalent carbon signals, which are characterized by one carbonyl

ketone (δ_C 218.2), two acetyls (δ_C 170.1 and 170.0), two sp² carbon (δ_C 158.9 and 118.0), seven methyl signals at (δ_C 27.5, 27.1, 21.4, 21.1, 21.0, 19.7 and 15.3), seven sp³ methylenes (including one oxygenated methylene type at δ_C 70.3), seven sp³ methynes (including two oxygenated types at δ_C 75.4, 75.6 and a hemiacetal at δ_C 99.8) and four quaternary carbons. These properties represented a total of four or nine degrees of unsaturation, and the remaining five degrees of unsaturation corresponded to limonoid framework. ^{16,17,22} Comprehensive analyses of the above mentioned 1H and ^{13}C NMR data exhibited that the structure of 4 was similar to that of neemfruitin, ²² a limonoid isolated from *Azadirachta indica*. Consequently, compound 4 was identified as neemfruitin A.

Compound 5 was obtained as a white solid with a proposed molecular formula of C28H40O5 based on HRTOFMS (m/z 455.0712 [M+H]⁺, (calcd. $C_{28}H_{40}O_5$ m/z456.2876) and NMR data (Table 1), therefore nine indices of hydrogen deficiency were required. Furthermore, the IR absorption bands at 3300, 1705 and 1080 cm⁻¹ show the presence of hydroxyl, carbonyl and ether properties, while the NMR data was highly similar to 4. However, differences were observed in the disappearance of one of acetyl group at $[\delta_H \ 1.91 \ (3H, s), \delta_C \ 21.7, \ 1701.1],$ oxygenated methyne at [δ_H 5.39 (1H, m), δ_C 75.4] and the appearance of olefinic signals at $[\delta_H 7.14 (1H, d, J=10.2)]$ Hz), 5.86 (1H, d, J = 10.2 Hz); δ_C 158.8, 125.7), suggesting 5 as a 1-deacetyl derivative of 4. The proton pairing was also confirmed with the ¹H-¹H COSY spectrum (Fig. 2) and supported the presence of limonoid structure in 5.22,23 The HMBC crosspeaks (Fig. 2) from H-2 ($\delta_{\rm H}$ 5.86) to C-1 $(\delta_C 158.8)$ and carbonyl at C-3 $(\delta_C 204.6)$, indicated the presence of α , β -unsaturated ketone at C-1, C-2 and C-3, respectively. Correlation which was arising from H-7 (δ_H 5.33) to C-1' (δ_C 170.1) and C-6 (δ_C 27.5) as well as from CH₃-2" (δ_H 1.94) to to C-1' (δ_C 170.1), indicate that acetyl group was attached at C-7. The above-mentioned NMR data suggested that compound 5 is a limonoid, for which NMR data (Table 1) showed many similarities to those of 23-desmethyllimocin-B, a limonoid isolated from the seed of *Azadirachta indica*.²³ Therefore, compound **5** was identified as 23-desmethyllimocin-B.

Compound **6** was obtained as a white solid with a molecular composition of C₃₂H₄₈O₆, based on HRTOFMS (*m/z* 527.2984 [M+H] +, (calcd. C₃₂H₄₈O₆ *m/z* 528.2951) and NMR analysis, therefore nine degrees of unsaturation are required. Furthermore, the IR spectrum showed the absorption bands for hydroxyl (3420 cm⁻¹), aliphatic (2949 and 2860 cm⁻¹), carbonyl (1715 cm⁻¹) and ether (1080 cm⁻¹) moieties, while the NMR data observed in

Table 2. IC₅₀ inhibition values of compounds **1 - 8** against MCF-7 breast cancer cell line

Compounds	IC ₅₀ (μM)					
Cabrealeolactone (1)	61.18					
Cabreadiol (2)	33.12					
Prototiamin A (3)	76.08					
Neemfruitins A (4)	181.12					
Desmethyllimocin B (5)	98.18					
Protoxylocarpin G (6)	90.24					
Melianodiol (7)	16.84					
Indicalilacol B (8)	20.23					
Cisplatin*	13.20					

^{*}positive control

Table 1 was highly similar to **5**. However, the difference was identified in the absence of a hemiacetal group at $[\delta_H 4.57 \text{ (1H, dd, } J=9.6, 1.8 \text{ Hz}), \delta_C 99.2]$, and also the appearance of a newly oxygenated methynes at $[\delta_H 4.47 \text{ (1H, dd, } J=9.6, 1.8 \text{ Hz}), 3.92 \text{ (1H, d, } J=9.6 \text{ Hz}), \delta_C 78.2, 74.4]$, two methyls at $[\delta_H 1.35 \text{ (3H, s)}, 1.44 \text{ (3H, s)}, \delta_C 26.9, 23.8]$ and one oxygenated quaternary carbon at $\delta_C 72.5$. These observations suggest that **6** is a 2-methylbutane-2,3-diol derivative of **5** and is attached to C-23. The aforementioned data as well as biogenetic considerations suggested that compound **6** is protolimonoid, with a structure similar to that of protoxylocarpin G, a protolimonoid isolated from *Xylocarpus granatum*. Consequently, compound **6** was identified as protoxylocarpin G.

Compound 7 was obtained as a colorless crystal. Its molecular composition C₃₀H₄₈O₅, was determined from HRTOFMS (m/z 487.4535 [M+H]⁺, (calcd. $C_{32}H_{50}O_7$ m/z488.4502) and NMR data (Table 1), therefore seven unsaturated degrees were required. The IR spectra showed absorption peaks at 3350 cm⁻¹ (OH), 2935 cm⁻¹ (C-H sp³), 1725 (C=O), 1645 (C=C), 1457 and 1380 cm⁻¹ (gem-dimethyl groups), and 1055 cm⁻¹ (C-O) groups. The ¹H-NMR spectrum showed the presence of seven tertiary methyl groups at δ_H 1.39 (CH₃-27), 1.32 (CH₃-26), 1.01 (CH₃-30), 0.93 (CH₃-28), 0.92 (CH₃-29), 0.84 (CH₃-18) and 0.78 (CH₃-19). One olefinic proton at $\delta_{\rm H}$ 5.26 (1H, dd, J=2.6, 5.8 Hz, H-7), two oxygenated methyne signals at $\delta_{\rm H}$ 3.64 (1H, ddd, J= 1.3, 6.2, 8.7 Hz, H-23), 4.06 (1H, br.s, H-24) and one hemiacetal signal at $\delta_{\rm H}$ 4.63 (1H, d, J = 5.6 Hz) were also observed in the ¹H-NMR spectrum. The ¹³C-NMR spectrum of 7 showed 30 carbons and were classified as seven methyl groups at δ_C 27.8, 27.4, 27.0, 26.5, 23.5, 21.6 and 13.1, one carbonyl ketone at $\delta_{\rm C}$ 218.4 (C-3), two oxymethine signals at $\delta_{\rm C}$ 86.4 (C-23) and 71.3 (C-24), one oxygenated quaternary

carbon at $\delta_{\rm C}$ 70.3 (C-24), hemiacetal signal at $\delta_{\rm C}$ 99.3 (C-21) and two sp² carbon at $\delta_{\rm C}$ 118.8 (C-7) and 145.3 (C-8), respectively. These properties represented two of seven total degrees of unsaturation, and the remaining five degrees of unsaturation were consistent with the limonoid structure. The above data showed general features similar to those of melianodiol, a limonoid isolated from the leaves of *Aglaia andamanica*. Therefore, compound 7 was identified as melianodiol.

Compound 8 was obtained as a white solid having a molecular composition of C₃₀H₄₈O₅, based on HRTOFMS $(m/z 487.4535 \text{ [M+H]}^+, \text{ (calcd. } C_{32}H_{50}O_7 \text{ } m/z 488.4502)$ along with NMR analysis, therefore seven degrees of unsaturation are required. Furthermore, the IR spectrum showed the absorption bands for hydroxyl (3430 cm⁻¹), aliphatic (2930), carbonyl (1760 cm⁻¹), gem-dimethyl (1420 and 1378 cm⁻¹) and ether (1080 cm⁻¹) moieties, while the NMR data observed in Table 1 was highly similar to 7. However, the difference was identified in the absence of a hemiacetal signal at $[\delta_H 4.63 (1H, d, J=5.6$ Hz), δ_C 99.3] and ketone signal at δ_C 218.4, and appearance of a newly carbonyl lactone at δ_C 178.3 (C-21) and hydroxyl signal at $[\delta_{\rm H} 4.32 \ (1 \, {\rm H}, \, {\rm d}, \, J = 5.4 \, {\rm Hz}), \, \delta_{\rm C} \, 72.4].$ These observations indicate that 8 is 3-hydroxy and lactone derivative of 7. The above data revealed that 8 is a tirucallane-type triterpenoid similar to indicalilacol B, isolated from Azadiracta indica.²⁶ Consequently, compound 8 was identified as indicalilacol B.

The isolated compounds, 1 - 8, were evaluated for their cytotoxic against MCF-7 breast cancer lines using a previously described method, 17,18,27 using cis-Platin as a positive control.^{28,29} Among the compounds tested, melianodiol (7), exhibited the strongest cytotoxicity with an IC₅₀ value of 16.84 μM. Interestingly, Compounds 7 and 8 share the same structure and only differed at the functional of the cyclopentane ring, the cytotoxic activities of compounds 7 and 8 differed greatly, suggested the presence of hemiacetal group increase the cytotoxic activity. These results were supported from previously studies that the presence of a hemiacetal group in triterpenoids structure can increase cytotoxic activity. 17,24,25 In addition, despite close structural similarities to 7, a low activity was observed in 6, which indicates the possible contribution of an acetyl group to cytotoxicity.25 This results was supported from previously studies that the presence of a hemiacetal group in triterpenoids structure can increase cytotoxic activity.²⁵ Therefore, the structural comparison of compound 7 with the other compounds indicates that carbonyl, hydroxyl and acetyl groups present contribute to the cytotoxic activity.

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